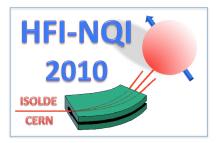
## **HFI/NQI 2010**



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## Structural Phase Transitions in a [Chloranilic Acid]-[2-Pyrrolidone] (1:2) Molecular Complex with Hydrogen-Bond Networks

Chloranilic acid has been widely studied as a proton donor forming hydrogen-bonds with various proton acceptors. Obtained complexes have been shown to have marked properties such as dynamic proton transfers, ferroelectricity, and so on. In the present study, we prepared a new chloranilic acid complex with 2pyrrolidone in search for new solid properties by measurements of single crystal X-ray diffraction (SCXRD), <sup>35</sup>Cl NQR, <sup>1</sup>H NMR, and differential scanning calorimetry (DSC). Crystals of [chloranilic acid]-[2-pyrrolidone] (1:2) complex were obtained by slow cooling of an acetonitrile solution containing 2-pyrrolidone (pyrr) and chloranilic acid (H<sub>2</sub>ca). The crystals were identified by elemental analysis and SCXRD. In the structure, it was found that pyrr and H<sub>2</sub>ca molecules are connected by hydrogen-bonds and make up a one-dimensional chain structure with hydrogen-bond networks. The onedimensional chains are stacked by  $\pi$ - $\pi$  interactions in the crystals. On [H<sub>2</sub>ca][pyrr]<sub>2</sub> crystals, we measured a temperature dependent <sup>35</sup>Cl NQR lines and spin-lattice relaxation times by using a Bruker BioSpin AVANCEII system. Two <sup>35</sup>Cl resonance lines were observed at room temperature (<i>Phase I</i>) in agreement with the SCXRD results. New four broad lines appear below 200 K (<i>Phase II</i>), where room-temperature two lines still persist with a gradually decrease of intensities upon cooling down to 175 K. Below 175 K (<i>Phase III</i>), another two lines with equal intensities appeared in the high frequency side. When heated from 77 K, these two peaks were repeated up to ca. 200 K where only the low-frequency peak disappeared. The high frequency peak survived with no anomaly at 200 K up to room temperature (<i>Phase Ia</i>), where the structure was shown to be different from the original one. It is expected that Phase <i>Ia</i> is stable and Phase <i>I</i> is a metastable phase. We believe this transition is due to an incommensurate structure.

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poster

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