

# Dense room temperature polarised nuclear targets using SABRE chemical hyperpolarisation – R&D status

Joint APP, HEPP and NP Conference - April 2024

**Benjamin Collins** <sup>A,B</sup>

Prof Dan Watts <sup>A</sup>

Dr Nick Zachariou <sup>A</sup>

Dr Mikhail Bashkanov <sup>A</sup>

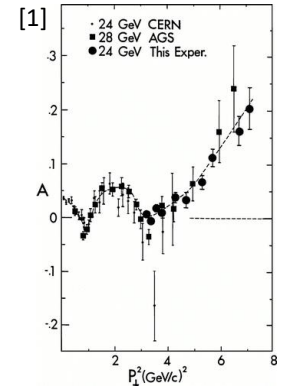
Prof Simon Duckett <sup>B</sup>

A - Department of Physics, Engineering and Technology, University of York, Heslington YO10 5DD, U.K.

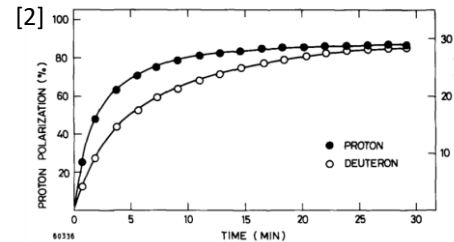
B - Centre for Hyperpolarisation in Magnetic Resonance, University of York, Heslington YO10 5NY, U.K.

# Polarised nuclear targets

- Help to answer questions such as:  
*“How do quarks and gluons carry the spin of protons?”*
- Current targets use frozen ammonia/butanol polarized by DNP.
- Cannot keep up with increasing beam intensities.
  - Strong depolarisation effects from heat deposition.
  - Long polarisation build-up times.
- We hope SABRE can resolve these issues.



Proton analysing power



DNP polarisation build-up



MAMI Frozen spin target

[1] G. Crabb and W. Meyer, Annual Review of Nuclear and Particle Science, vol. 47, no. 1, pp. 67–109, Dec. 1997.

[2] de Boer W. CERN Yellow Report 74-11 (1974); J. of Low-Temp. Phys. 22:185 (1976).

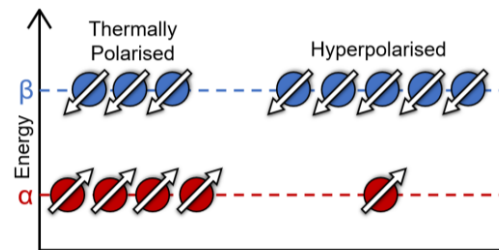
[3] Thomas, A. et. al. Phys. Part. Nuclei 44, 964–967 (2013).

# Hyperpolarisation and *parahydrogen*

Hyperpolarisation → greater than equilibrium spin polarisation, given by:

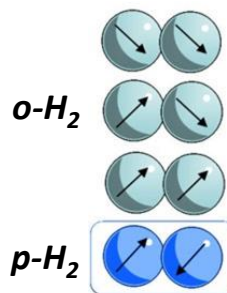
$$P = \frac{p_\alpha - p_\beta}{p_\alpha + p_\beta}$$

Thermal equilibrium polarisation:  $P_{eq} = \tanh\left(\frac{\hbar\gamma B_0}{2k_B T}\right) \ll 1$

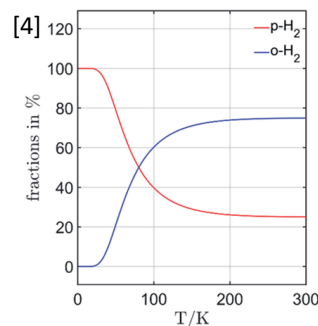


Parahydrogen (*p*-H<sub>2</sub>) → Singlet nuclear spin isomer of molecular hydrogen (H<sub>2</sub>).

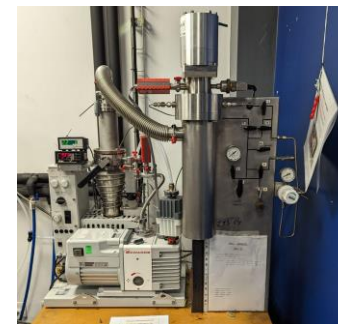
Readily produced by passing H<sub>2</sub> over a ferromagnetic catalyst at low temperatures.



Spin configurations of H<sub>2</sub>.



*p*-H<sub>2</sub> % by temperature.

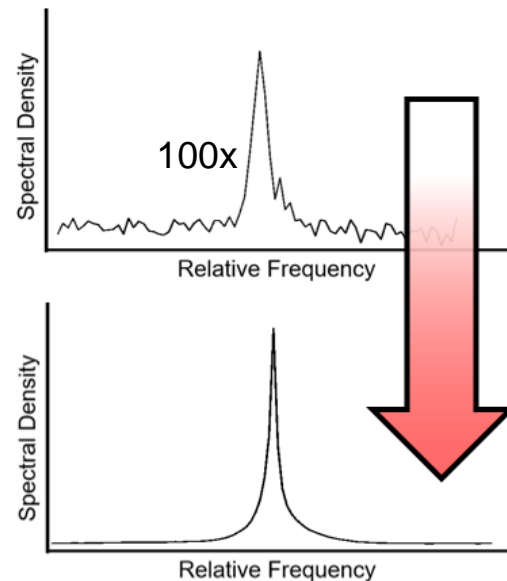
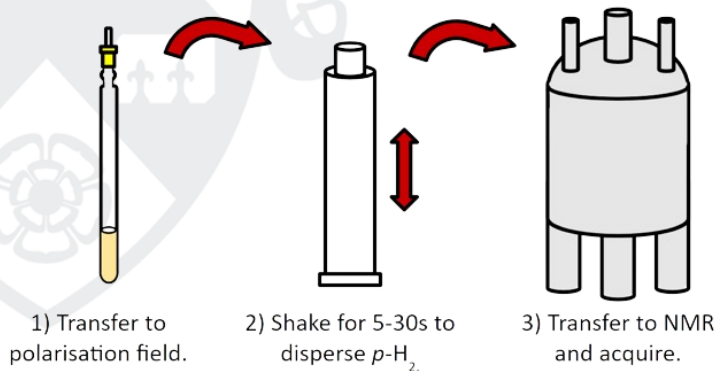


*p*-H<sub>2</sub> generator at CHyM.

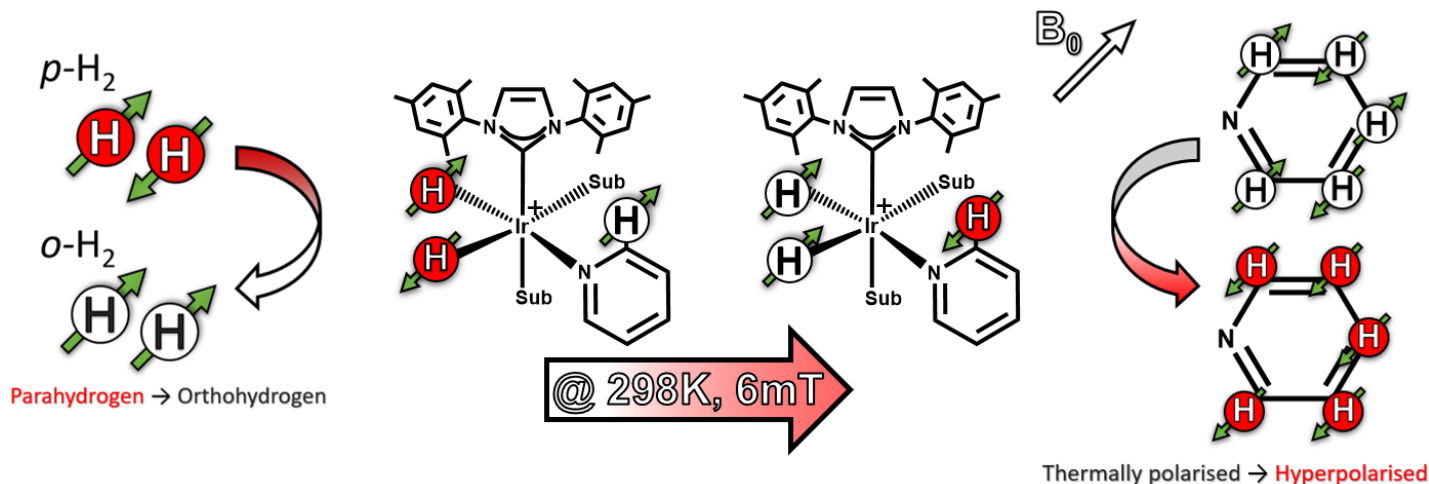
# SABRE

- Chemically-catalysed nuclear hyperpolarisation technique.
- Uses  $p\text{-H}_2$  feedstock of spin order.
- Works at **room temperature** in **weak magnetic fields** ( $\mu\text{T}$ - $\text{mT}$ ).
- Polarisation can be **generated continuously**.
- $^1\text{H}$ ,  $^{13}\text{C}$ ,  $^{15}\text{N}$  (+ more).

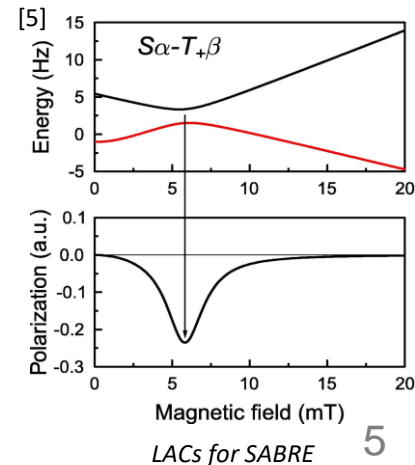
SABRE 'Shake and drop' acquisition.



# SABRE spin order transfer

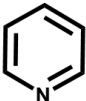
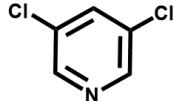
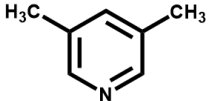
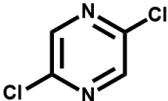


- SABRE substrates and  $p\text{-H}_2$  brought together by SABRE catalyst.
- Spin exchange occurs through  $J_{\text{HH}}$  couplings.
- LACs for an allow the transition  $S\alpha \rightarrow T_+\beta$  at specific magnetic fields.
- Polarisation reaches a maximum within seconds.



# Substrate comparison

Data from selected substrates under equivalent conditions.

Pyridine	3,5-Dichloropyridine	3,5-Dimethylpyridine	2,5-Dichloropyridazine
$C_5H_5N$	$C_5H_3Cl_2N$	$C_7H_9N$	$C_4H_2Cl_2N_2$
Polarisable proportion of protons (%)			
11.9	4.1	15.5	2.6
Polarisation yield (%)			
1.4	2.9	0.15	Low (N/A)
Avg $T_1$ (s)			
40.0	83.8	12.2	128.5
			

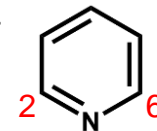
Trends seen:

+ Methyl group: **Polarisable protons** ↑, **Polarisation lifetime** ↓.

+ Halogen: **Polarisable protons** ↓, **Polarisation lifetime** ↑.

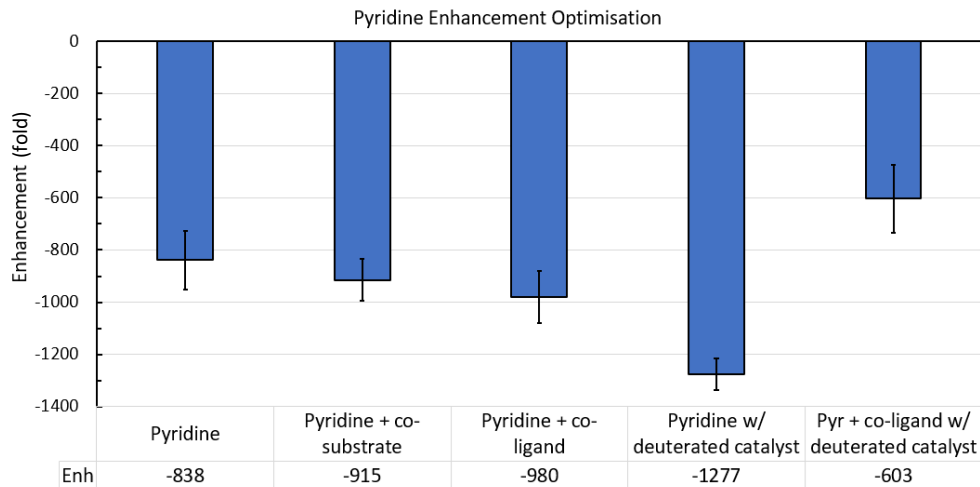
→ Addition of low- $\gamma$  nuclei can reduce relaxation in hyperpolarised material.

→ Addition of halide/methyl groups can reduce binding efficiency, especially in positions 2 & 6.

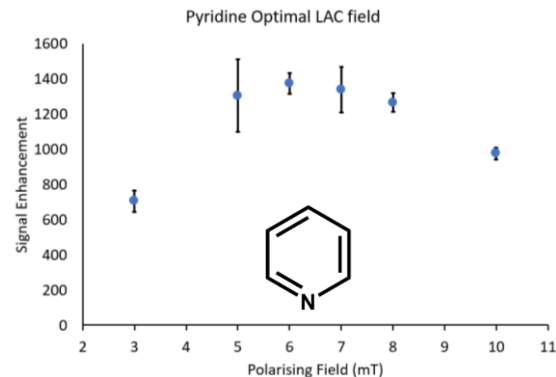
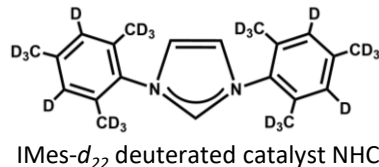
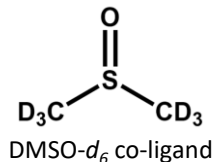
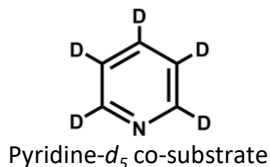


# Pyridine Enhancement Optimisation

Substrate of interest due to high polarizable proton fraction and high yields.



- + Co-substrate: Enh ↑
- + Co-ligand: Enh ↑↑
- + Deuterated catalyst: Enh ↑↑↑
- + Co-ligand and deuterated catalyst: Enh ↓

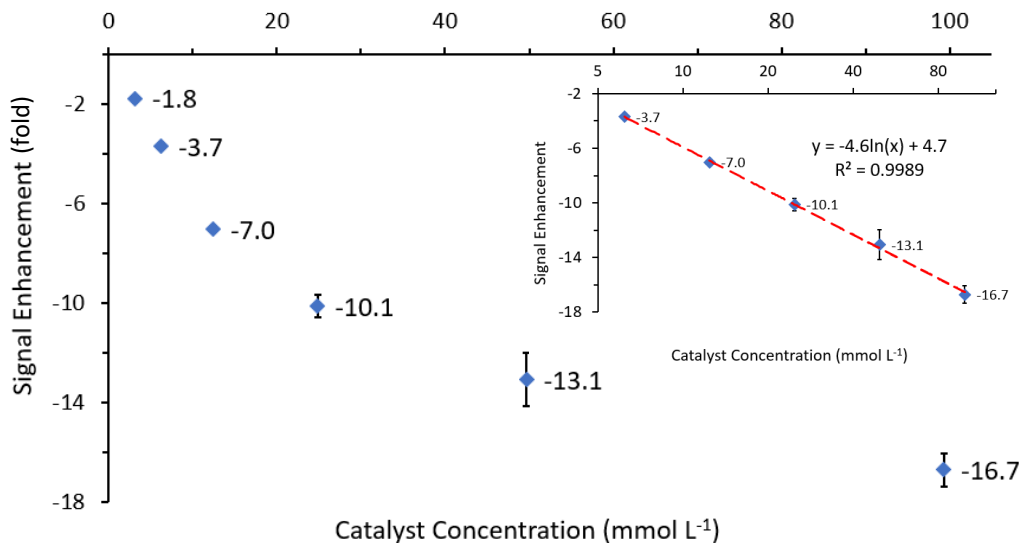


→ Optimum field for LAC found to be 6 mT for  $^1\text{H}$ .

# Limitations of $p\text{-H}_2$ solubility

SABRE typically performed in solution using  $\text{MeOD-}d_4/\text{DCM-}d_2$  as a solvent.  
What happens if SABRE is performed on a neat solution of substrate?

Signal Enhancement vs Catalyst Concentration for 1% Pyridine, 99% Pyridine- $d_5$



- Signal enhancement (Enh) seen is  $10^2\text{-}10^3$  fold lower than for optimal conditions.
  - Enh doesn't scale linearly with catalyst concentration - dependence is logarithmic.
  - Solubility of  $p\text{-H}_2$  in solution is the limiting factor.
- Polarization of neat liquids via SABRE is not feasible without extreme pressures.

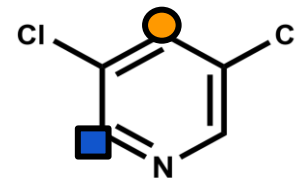


# Relaxation studies on model substrate



Relaxation decay constant,  $T_1$ , is a key metric in hyperpolarisation:

- Determines the maximum polarisation levels that can be reached.
- Long  $T_1$  reduces the need for high  $p\text{-H}_2$  pressures and catalyst concentrations.



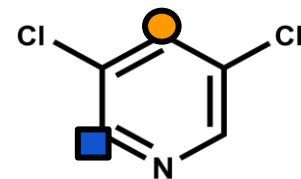
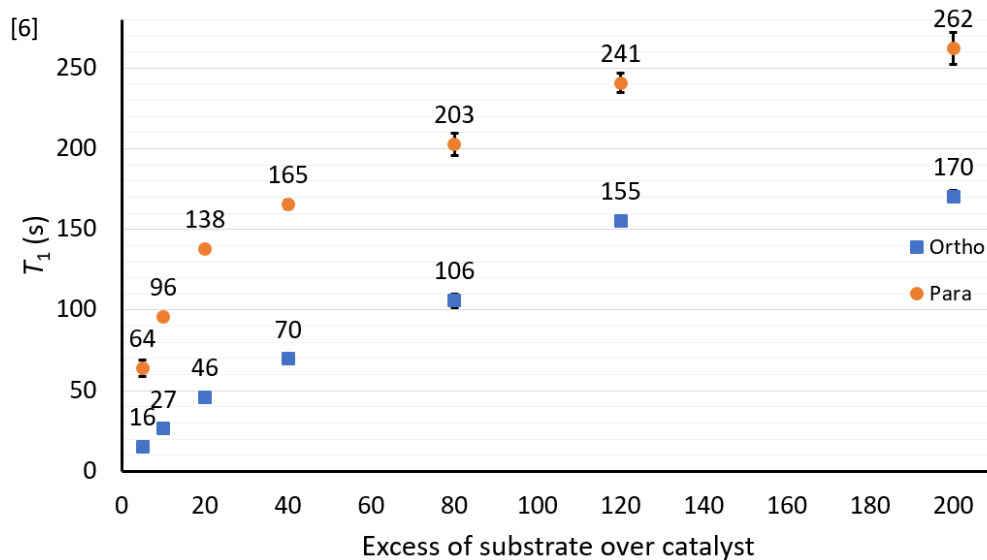
Model substrate -  
3,5-dichloropyridine

Conclusions:

- a) Relative catalyst concentration in solution is the primary driver of relaxation.
- b) Dependence of relaxation on holding field is small in the range 1.4-11.7T.
- c) Relaxation  $T_1$  peaks at 30°C.

# Relaxation studies on model substrate

a) Variation of  $T_1$  with substrate excess



Model substrate -  
3,5-dichloropyridine

Conclusions:

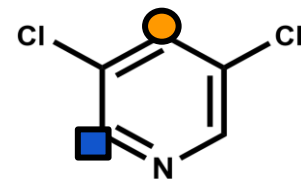
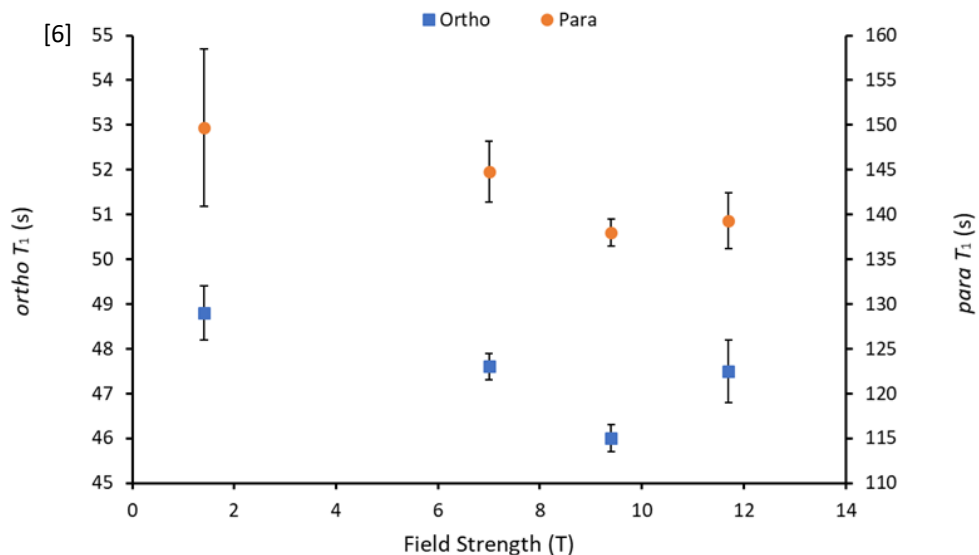
a) Relative catalyst concentration in solution is the primary driver of relaxation.

b) Dependence of relaxation on holding field is small in the range 1.4-11.7T.

c) Relaxation  $T_1$  peaks at 30°C.

# Relaxation studies on model substrate

b) Variation of  $T_1$  with holding field



Model substrate -  
3,5-dichloropyridine

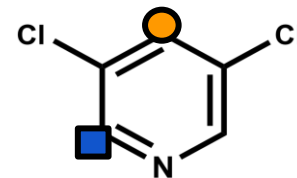
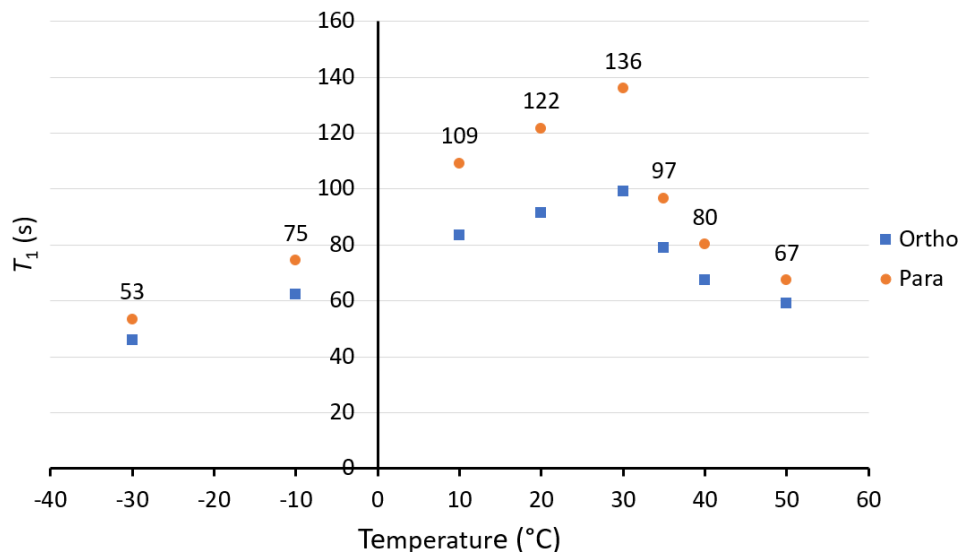
Conclusions:

- Relative catalyst concentration in solution is the primary driver of relaxation.
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c) Relaxation  $T_1$  peaks at 30°C.

# Relaxation studies on model substrate

c) Variation of  $T_1$  with temperature



Model substrate -  
3,5-dichloropyridine

## Conclusions:

- Relative catalyst concentration in solution is the primary driver of relaxation.
- Dependence of relaxation on holding field is small in the range 1.4-11.7 T.
- Relaxation  $T_1$  peaks at 30°C.

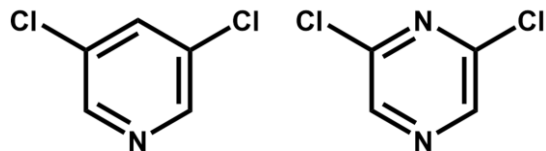
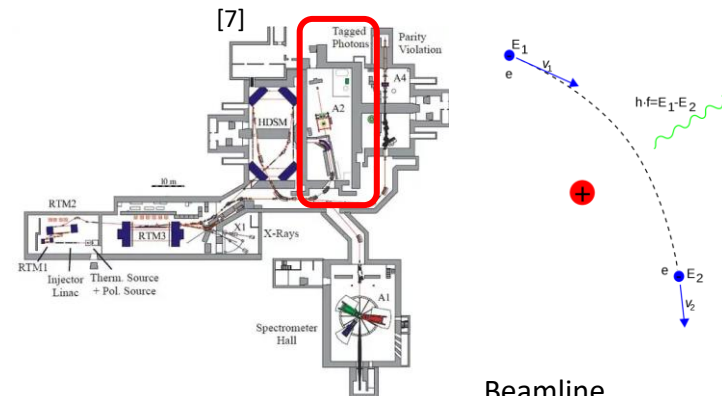
# Polarization resilience test @ MAMI



UNIVERSITY  
of York

Aim → Test resilience of SABRE hyperpolarization to an incident photon beam by monitoring rate of polarization decay.

- Facility produces energy-tagged Bremsstrahlung photons from a high energy electron beam.  $E = 40 \text{ MeV} \rightarrow 1.6 \text{ GeV}$ .
- 2 halogenated pyridine/pyrazines chosen for long polarisation lifetimes.
- Additional sample left in high dose area (electron beam dump) to investigate effects of radiation damage.
- Measurements made using commercially available, low cost benchtop MRI.

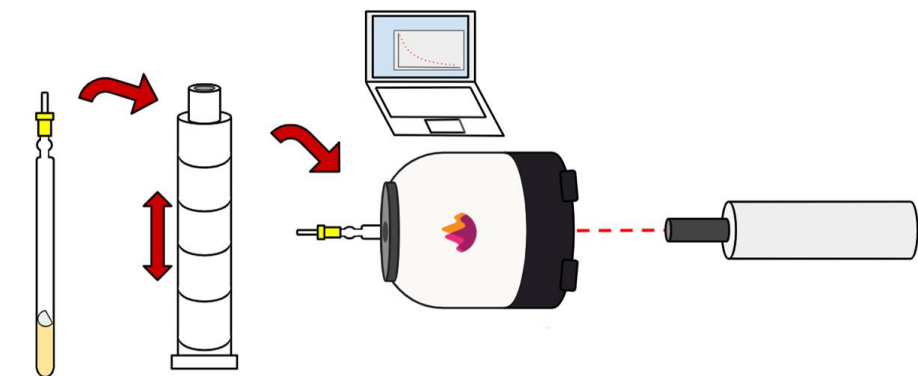


Test substrates - 3,5-dichloropyridine (left), 3,5-dichloropyrazine (right).



# Experimental procedure and pulse sequence

## Experimental procedure:



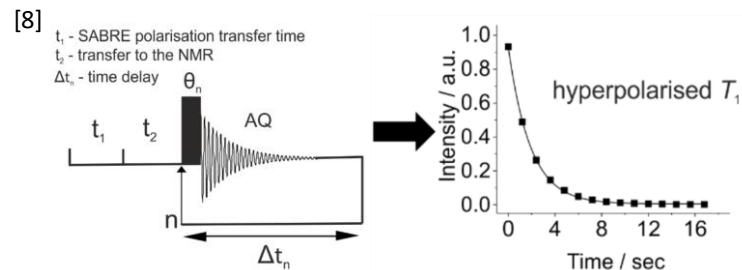
1) Prepare sample and pressurise to 5 bar.

2) Transfer to Halbach array & shake for 45s.

3) Transfer to MRI & start acquisition.

4) Vacate hall and turn on photon beam.

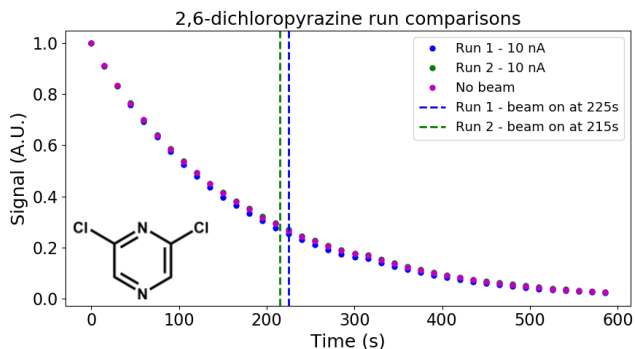
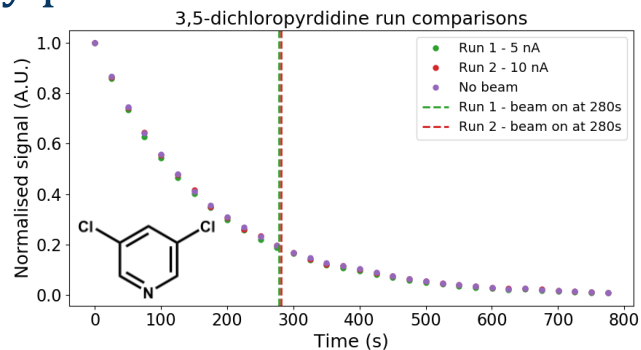
## Pulse sequence:



Variable flip angle pulse sequence uses many small flip angle RF pulses to sample the magnetisation.

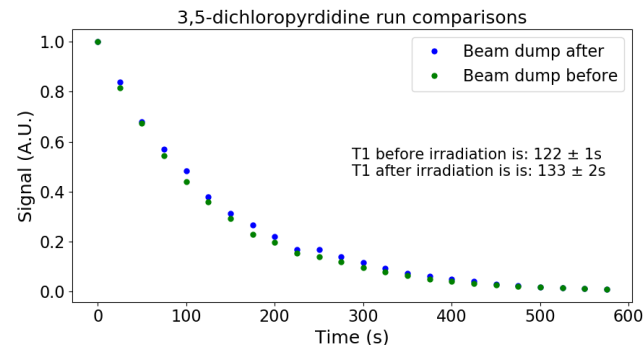
# Results from MAMI

## Decay plots



→ No visible change in relaxation rate for runs with/without beam.

## High dose sample



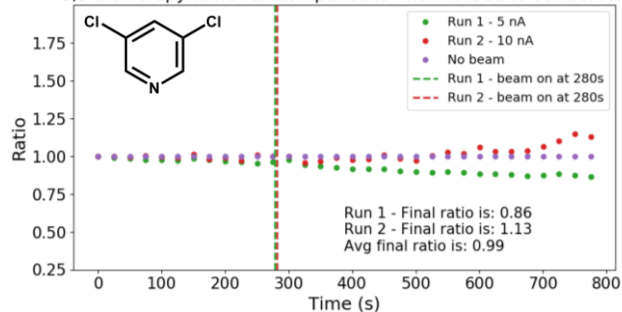
Irradiated in highest dose area of the facility - electron beam dump.

- $T_1$  increased by 9%.
- Polarisation yield decreased by 13%.
- No colour change.

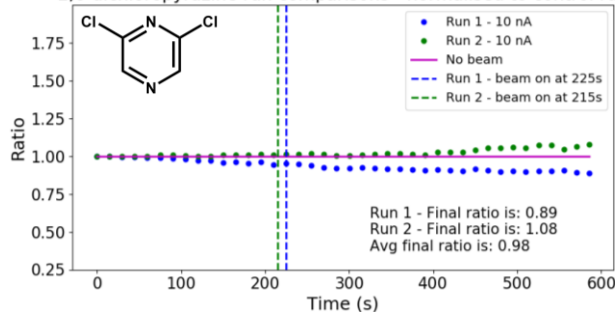
→ Polarisation catalyst not significantly affected.

# MAMI results continued

3,5-dichloropyridine run comparisons - normalised to control run



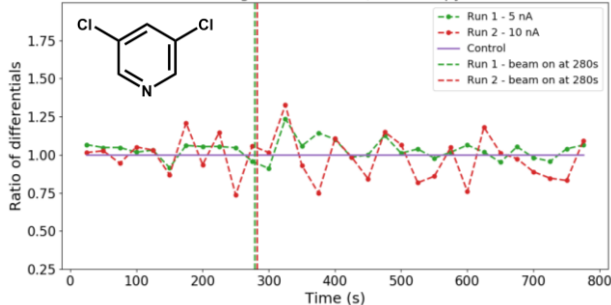
2,6-dichloropyridazine run comparisons - normalised to control run



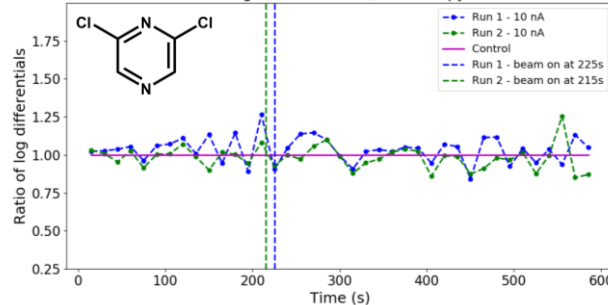
Ratio of polarized signal in beam-on and control measurements.

- Avg. ratio to control is close to 1.

Ratio of log differentials - 3,5-dichloropyridine



Ratio of log differentials - 2,6-dichloropyridazine



Ratio of log. gradients of the beam-on runs to the control run.

Ratio of 1 shows same rate of decay - ratio of 0.5/2 shows half/double rate of decay.

→ No evidence for significant levels of beam-induced decay.

Run	Mean ratio before beam	Mean ratio after beam
Run 1	1.02±0.05	1.04±0.07
Run 2	1.00±0.02	0.97±0.15
Avg	1.01±0.07	1.01±0.10

Run	Mean ratio before beam	Mean ratio after beam
Run 1	1.04±0.06	1.03±0.09
Run 2	0.99±0.05	0.98±0.09
Avg	1.02±0.04	1.0±0.07



# Summary



SABRE is a **cost-effective** and easy to implement nuclear polarization technique which operates at **room temperature** in **weak fields** and may be able to overcome some of the issues facing DNP polarized targets.

→ Comprehensive substrate scope performed with **key substrates of interest identified**.

→ Optimisation of polarization yields undertaken with **>10% <sup>1</sup>H polarization achieved**.

→ Relaxation studies performed to identify optimum running conditions.

→ First measurements on the **resilience of SABRE polarization** to an incident photon beam & effects of high accumulated dose.

→ Showed that polarization monitoring can be performed accurately on a **commercially available, low-cost** benchtop MRI.

# Acknowledgements

Thanks to Dan Watts, Mikhail Bashkanov, Nick Zachariou and Simon Duckett for their help and guidance with the project as well as everyone in Medical Physics and CHyM.



CHyM Dec '23



Nuclear Physics Dec '23



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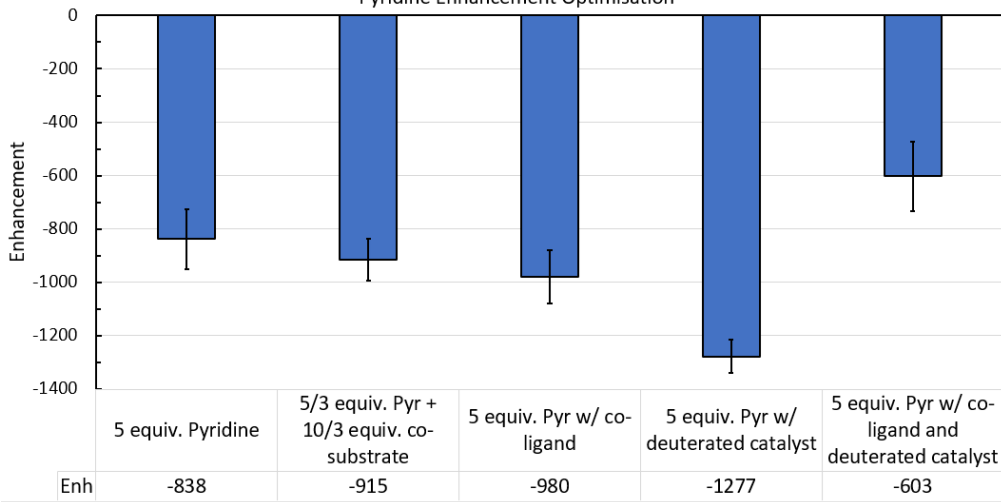
# Supplementary Slides

Joint APP, HEPP and NP Conference - April 2024

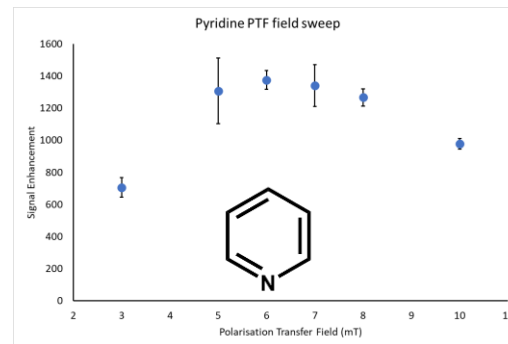
# Pyridine Enhancement Optimisation

Substrate of interest due to high polarizable proton fraction and high yields.

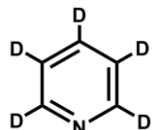
Pyridine Enhancement Optimisation



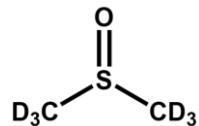
- + Deuterated co-substrate: Enh ↑
- + Co-ligand: Enh ↑↑
- + Deuterated catalyst: Enh ↑↑↑
- + Co-ligand and deuterated catalyst: Enh ↓



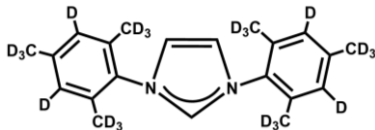
→ Optimum PTF found to be 6 mT for  $^1\text{H}$ .



Pyridine- $\text{d}_5$  co-substrate




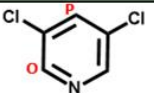

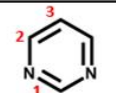
DMSO- $\text{d}_6$  co-ligand


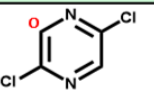
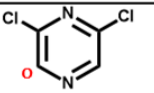
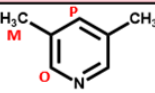


IMes- $\text{d}_{22}$  deuterated catalyst NHC

# Substrate comparison

Data from selected substrates under equal conditions (40 equiv. substrate, 5 equiv. DMSO-d6 co-ligand in DCM-d2).

Pyridine	3,5-Dichloropyridine	Pyrazine	Pyrimidine
C <sub>5</sub> H <sub>5</sub> N	C <sub>5</sub> H <sub>3</sub> Cl <sub>2</sub> N	C <sub>4</sub> H <sub>4</sub> N <sub>2</sub>	C <sub>4</sub> H <sub>4</sub> N <sub>2</sub>
Polarisable proportion of protons (%)			
11.9	4.1	9.5	9.5
Average polarisation yield (%) [Total]			
1.37±0.08	2.87 ± 0.02	1.82 ± 0.05	0.54 ± 0.09
T1 relaxation const. (s)			
O/M/P = 41.2/39.5/37.0	O/P = 69.4 / 112.6	O = 42.0	1/2/3 = 69.5/49.8/37.8
			

Pyridazine	2,5-Dichloropyrazine	2,6-Dichloropyrazine	3,5-Dimethylpyridine
C <sub>4</sub> H <sub>4</sub> N <sub>2</sub>	C <sub>4</sub> H <sub>2</sub> Cl <sub>2</sub> N <sub>2</sub>	C <sub>4</sub> H <sub>2</sub> Cl <sub>2</sub> N <sub>2</sub>	C <sub>7</sub> H <sub>9</sub> N
Polarisable proportion of protons (%)			
9.5	2.7	2.7	15.5
Average polarisation yield (%)			
0.49±0.06	N/A (Low)	0.86±0.03	~0.15
T1 relaxation const. (s)			
O/M = 37.7/41.2	O = 128.5	O = 61.8	O/M/P = 16.9/8.1/27.0
			

Trends seen:

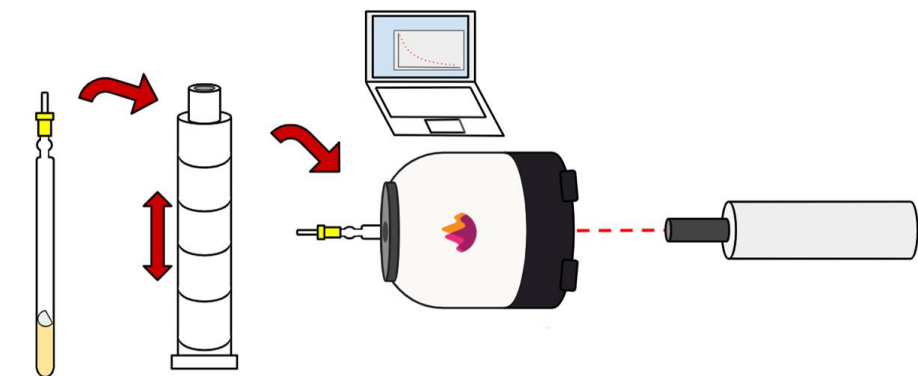
- + Methyl group: Polarizable protons ↑, Polarization lifetime ↓
- + Halogen: Polarizable protons ↓, Polarization lifetime ↑

→ Addition of low- $\gamma$  nuclei can reduce relaxation in hyperpolarized material.

→ Addition of halide/methyl groups can cause steric hindrance, especially in positions 2 & 6.

# Experimental procedure and pulse sequence:

## Experimental procedure:



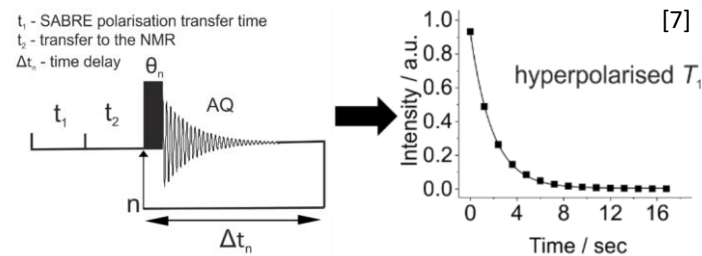
1) Prepare sample and pressurise to 5 bar.

2) Transfer to Halbach array & shake for 45s.

3) Transfer to MRI & start acquisition.

4) Vacate hall and turn on photon beam.

## Pulse sequence:



Variable flip angle pulse sequence uses small tip angle RF pulses to sample the magnetisation:

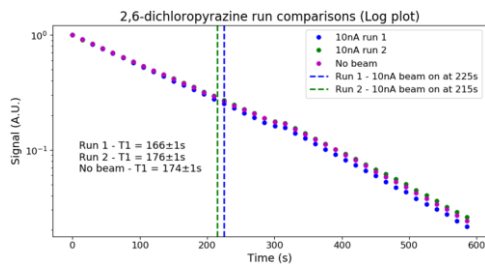
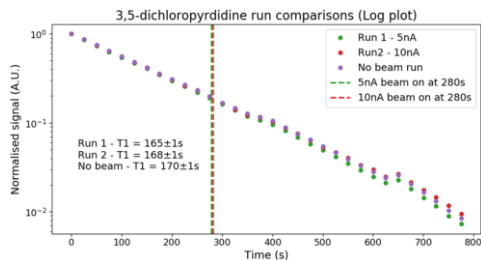
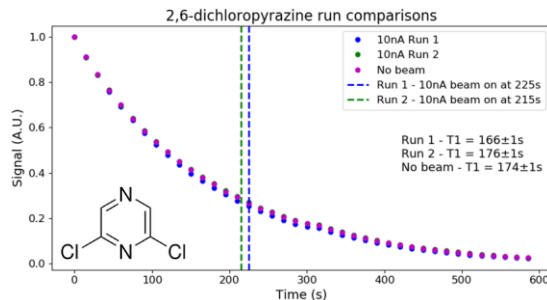
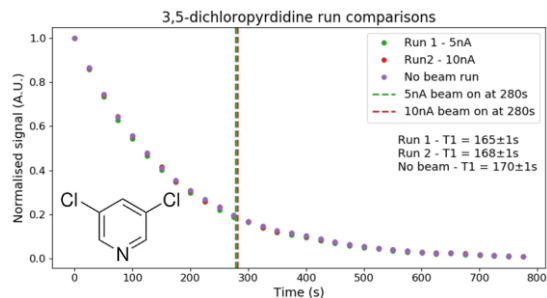
$$M_{z,n} = M_{z,n-1} \exp\left(\frac{-\Delta t_{n-1}}{T_1}\right) \cos \theta_n$$

$$M_{xy,n} = M_{z,n} \sin \theta_n$$

$M_{z,n}$  and  $M_{xy,n}$  are the longitudinal and transverse magnetisation,  $\Delta t_n$  is the delay following, and  $\theta_n$  is the tip angle used for the  $n^{\text{th}}$  scan.

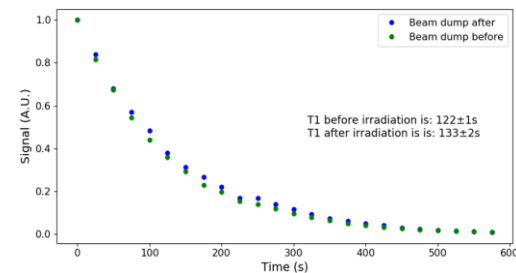
# Results

## Decay plots



→ No visible change in relaxation rate for runs with/without beam.

## High dose sample



Irradiated in highest dose area of the facility - electron beam dump.

- T1 increased by 9%.
- Polarisation yield decreased by 13%
- No colour change.

→ Polarisation catalyst not significantly affected.